

# Simultaneous estimation of Losartan and Atenolol by UV Spectrophotometric Method

A. Gajbhiye and N. Dwivedi

Department of Pharmaceutical Sciences, Dr. H. S. Gour Central University, Sagar (M.P.) 470003 India, Email ID : asmitapatil27@rediffmail.com

Abstract - Losar-beta is available for the treatment of hypertension. It contains losartan potassium (LS; 50mg) and atenolol (AT; 50mg). In the present study, simple, rapid, precise and accurate method for the simultaneous estimation of these drugs have been developed and validated by UV spectrophotometry. The method was validated with respect to its linearity, limit of quantitation (LOQ), limit of detection (LOD), precision and accuracy. In this method the LS and AT were scanned with water: methanol (80:20) as solvent and  $\lambda$ max were found to be at 232 and 275 nm for LS and AT respectively. For LS  $(A_1 = 0.0054 \text{ Cx} + 0.0634)$ Cy) and AT  $(A_2 = 0.0028Cx + 0.0128Cy)$ . The equations were developed by Vierodt's method. The LOD was found to be 0.380 µg/mL for LS and 0.860 µg/mL for AT. LOQ was 1.160 µg/mL for LS and 2.600 µg/mL for AT. The %RSD for day to day precision was found to be 0.0060 for LS and 0.0200 for AT. Percentage recovery was found to be  $99.32 \pm 0.08$ for LS and 99.54± 0.12 for AT. The linearity was found to be in the concentration range of 5-50 µg/mL for LS and AT. Statistical analysis proves that, the method is repeatable and selective for the analysis of LS and AT. The result of recovery studies for tablet was found to be nearly 100% showing no interference due to excipients. This method for simultaneous estimation of LS and AT is quite accurate, precise, economic, simple and rapid, hence can be employed for routine analysis in quality control laboratories.

Keyword – losartan, atenolol, UV spectrophotometry, method validation, simultaneous estimation.

#### **1. INTRODUCTION**

Losar-beta is available for the treatment of hypertension. It contains losartan potassium (LS; 50mg) and atenolol (AT; 50mg). Losartan is a agiotensin II receptor antagonist and atenolol is  $\beta$ 1 receptor antagonist. These drugs are more effective in combination therapy as compared to monotherapy [1, 2]. The combinations of these drugs are marketed under various brand names as tablet dosage form. Literature reveals that very few spectrophotometric methods are available for the simultaneous estimation for these combinations, which are expensive also. Hence, it was thought that a simultaneous estimation for these combinations can be carried out to make the methods more cost effective [3, 4].

If samples contains two absorbing drugs x and y each of which has absorption maxima at  $\lambda_1$  and  $\lambda_2$ . It may be possible to determine both drugs by simultaneous equation [5, 6]. The following criteria may be applied.

The information required is:

The absorptivities of x at  $\lambda_1$  and  $\lambda_2$ , ax<sub>1</sub> and ax<sub>2</sub> respectively.

The absorptivities of y at  $\lambda_1$  and  $\lambda_2$ ,  $ay_1$  and  $ay_2$  respectively.

The absorbances of the diluted sample at  $\lambda_1$  and  $\lambda_2$ ,  $A_1$  and  $A_2$  respectively.

Let cx and cy be the concentrations of x and y respectively in the diluted sample.

Two equations are constructed based upon the fact that at  $\lambda_1$  and  $\lambda_2$  the absorbance of the mixture is the sum of the individual absorbances of x and y.

At  $\lambda_1$  A<sub>1</sub> = ax<sub>1</sub> bcx + ay<sub>1</sub> bcy

At  $\lambda_2$  A<sub>2</sub> = ax<sub>2</sub> bcx + ay<sub>2</sub> bcy

For the measurements in 1cm cells, b = 1

Atenolol



Losartan potassium



#### **2. EXPERIMENTAL**

#### **Preparation of calibration curve:**

The stock solutions of 10  $\mu$ g/mL concentration were prepared in methanol and water (20:80 v/v). For atenolol three absorption maxima were observed at value of 225 nm, 275nm and 322nm. The  $\lambda$ max 275 nm was used for this study. For losartan potassium an absorption maxima observed at 232nm was used.

Working standard solution (100  $\mu$ g/mL) was made from stock solution (S) by suitably diluting with methanol. Aliquots (0.5, 1.0, 1.5....... 5.0 mL) were taken from

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this working standard solution and suitably diluted with methanol to give a concentration range of 5 to 50  $\mu$ g/mL. For atenolol and losartan the absorbance were recorded at 275 nm and 232 nm respectively against a reagent blank and calibration curve was plotted as shown in Fig. 1 and 2.



Fig. 1: Calibration curve of atenolol at 275 nm



Fig. 2: Calibration curve of losartan potassium at 232 nm



Fig. 3: Overlay UV spectra of atenolol & losartan potassium

#### **Optical characteristics**

The optical characteristics such as absorption maxima, Beer's law limit, correlation coefficient (r), slope (m), intercept (c), molar absorptivity and Sandell's sensitivity were calculated and the results are shown in Table 1. The absorption coefficient data of atenolol and losartan are given in Table 2 and 3 respectively.

Table 1:	Optical	characteristics
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Description	Values			
Parameters	Atenolol	Losartan potassium		
$\lambda_{max}$ (nm)	275	232		
Beer's Law	5-60	5-70		
Limit (µg/mL)				
Molar	$0.0150 \times 10^4$	0.2681×10 <sup>4</sup>		
absorptivity	0101200110			
(L/mol.cm)*				
Sandell's	47.25	6.67		
sensitivity (µg				
$cm^{-2}/(0.001)*$				
Regression	y=0.0256x+0.0073	y=0.283x+0.1365		
equation (y =				
mx + c)				
Slope (m)	0.0256	0.283		
Intercept (c)	0.0073	0.1365		
Correlation	0.9942	0.995		
coefficient (r)				

Table 2: Absorption coefficient data of atenolol

Conc.	275 1	nm	232	nm
(µg/mL)	Abs.	$E_{1cm}^{1\%}$	Abs.	$E_{1cm}^{1\%}$
5	0.0387	7.74	0.0151	3.02
10	0.0635	6.35	0.0247	2.47
15	0.0712	4.75	0.0341	2.27
20	0.1132	5.66	0.0503	2.51
25	0.1355	5.42	0.0675	2.70
30	0.1549	5.16	0.0676	2.25
35	0.1818	5.19	0.0725	2.071
40	0.2158	5.39	0.0839	2.098
45	0.2396	5.32	0.0966	2.14
50	0.2654	5.31	0.1957	3.91
	(N	(Iean)	(M	ean)
	a <sub>x1</sub>	= 5.393	a <sub>x2</sub> =	= 2.816

#### **Development of simultaneous equation**

For atenolol and losartan potassium Vierodt's equations were developed for simultaneous estimation by using the following set of equations:

At 275nm A1 = ax1 bCx + ay1 bCy

At 232nm A2 = ax2 bCx + ay2 bCy

Cx and Cy = Concentration of losartan potassium and atenolol respectively in  $\mu g$  /mL.

A1 and A2 = absorbance at 275 nm and 232 nm respectively.

ax1 and ax2 = absorption coefficient of losartan potassium at 275 nm and 232 nm respectively.

ay1 and ay2 = absorption coefficient of atenolol at 275 nm and 232 nm respectively.

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b = 1 (for measurement in 1 cm. Cells). Substituting the values of ax1, ax2, ay1 and ay2 equation can be rearranged as:

> $A_1 = 0.0054 Cx + 0.0634 Cy$  $A_2 = 0.0028 Cx + 0.012 Cy$

 
 Table 3: Absorption coefficient data of losartan potassium

Concentration	232 nm		275 nm	
(µg/mL)	Abs.	$E_{1cm}^{1\%}$	Abs.	$E_{\scriptscriptstyle 1cm}^{ m 1\%}$
5	0.3678	68.84	0.0687	13.74
10	0.5425	66.78	0.1161	11.61
15	0.6687	64.85	0.1765	11.76
20	0.8396	66.28	0.2207	11.04
25	1.0856	64.84	0.2914	11.66
30	1.3417	62.75	0.3593	11.98
35	1.7672	62.46	0.3855	11.01
40	1.9890	60.67	0.4637	11.59
45	2.1351	59.10	0.5105	11.34
50	2.3005	57.36	0.5627	11.25
	(Mean) a	ι <sub>Υ1</sub> =	(Mean) a	ι <sub>Y2</sub> =
	63.39		12.80	

# **3. RESULTS**

## **Estimation from tablets**

The powdered tablets (powder equivalent to 50 mg atenolol and 50 mg losartan potassium) were taken in 100 mL of conical flasks separately. These were extracted with 4 X 20 mL portion of water and methanol (80:20), and filtrate was taken in 100 mL volumetric flasks and the volumes were made up to 100 mL with water and methanol (80:20). Aliquots of a definite concentration were further suitably diluted to give the concentration in the range of 5-50  $\mu$ g/mL. The drug content in the tablets was calculated [5, 6, 7]. The experiments were repeated six times to check its reproducibility and the results are shown in Table 4, 5 and 6.

# **Method Validation**

#### Accuracy

The % recovery was found to be  $99.54\pm0.12$  for atenolol and % recovery was found to be  $99.32\pm0.08$  for losartan . Results of % recovery found are shown in Table 7 and 8.

## Linearity

It was linear in the range of  $5\mu g/mL - 50\mu g/mL$  for both atenolol and losartan potassium at 275 and 232 nm respectively.

Table 4: Estimation of drugs from Losar-beta tablet

Abs. Conc. found (µg/mL)	Amt. found/ tablet (mg)	Amt. Claimed/ tablet(mg)
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# Current Trends in Technology and Science ISSN: 2279–0535. Volume : 1, Issue : 3

275 nm	232 nm	AT (C <sub>X</sub> )	LS (C <sub>Y</sub> )	X	Y	X	Y
0.012	0.019	9.97	10.01	49.88	50.05	50.0	50.0
0.012	0.019	10.0	10.0	50.04	50.03	50.0	50.0
0.012	0.019	10.0	9.98	50.04	49.9	50.0	50.0
0.012	0.019	10.0	9.97	50.03	49.882	50.0	50.0
0.012	0.019	10.0	10.0	50.04	50.04	50.0	50.0
0.012	0.019	9.99	9.99	49.98	49.960	50.0	50.0

Table 5: Statistical results of Losar-beta tablets

Parameters	Atenolol	Losartan	
Standard	0.0671	0.0328	
Coefficient of	0.0997	0.0007	
variation Standard error of	0.0274	0.0134	
mean Percentage	50 1305 +	<i>4</i> 9 98 +	
range of error	0.0537	0.0262	
confidence			
limits)			

 Table 6: Compilations of results of statistical analysis

 of commercial formulations of Atenolol and losartan

potassium						
Comp- onent	Label claim	Amt. found	S.D.	% RSD	SE	
AT	50	$50.13 \pm 0.054$	0.0671	0.1338	0.0274	
LS	50	$49.98 \pm 0.026$	0.0328	0.0659	0.0134	

Values represents average of six determinations

Table 7: R	esults of	drug	recovery	studies	of	atenolol
T 1 P						

Level of standard addition (%)	Amour	% recovery ± SD		
90	49.82	49.98	49.86	99.88 ± 0.12
100	49.48	49.78	49.88	$99.34 \pm 0.28$
110	49.28	49.98	49.94	99.65 ± 0.32

 Table 8: Results of drug recovery studies of losartan

 potassium

Level of standard addition (%) Amount found in replicate	three % recovery ± SD
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#### Precision

# Intermediate precision: day to day

The % RSD for a tenolol was 0.0200 and for losartan potassium it was 0.0060

 Table 9: Intermediate precision: day-to-day

Atenolol			Losartan		
Conc. (µg/mL)	Day-1	Day-2	Conc. (µg/mL)	Day-1	Day-2
5	50.28	50.18	5	50.22	50.42
10	50.18	50.12	10	49.86	50.28
20	50.02	49.98	20	50.06	49.89
Mean	50.16	50.093	Mean	50.05	50.20
Mea	an	50.13	Mea	an	50.12
S.I	).	0.05	S.E	).	0.11
%R	SD	0.094	%R	SD	0.212

## Intermediate precision: analyst to analyst

The % RSD for atenolol 0.0036 and for losartan potassium it was 0.0034

### Table 10: Intermediate precision: analyst-to-analyst

Atenolol			Losartan		
Conc. (µg/mL)	A-1	A-2	Conc. (µg/mL)	A-1	A-2
5	50.32	50.12	5	50.18	50.28
10	50.10	49.78	10	49.82	49.84
20	50.20	50.26	20	50.08	49.98
Mean	50.21	50.05	Mean	50.02	50.03
Mean		50.13	Mean		50.03
S.D.		0.100	S.D.		0.002
%	RSD	0.100	%RSD		0.003

A-1 = Analyst1, A-2 = Analyst 2 Limit of Detection (LOD)

Based on the standard deviation of the response and slope

The detection limit may expressed as:

 $LOD=3.3\;\sigma\,/\,S$ 

Where  $\sigma$  = the standard deviation of the response S = the slope of the calibration curve.

For a tenolol LOD was found to be 0.860  $\mu g/mL$  and for losartan potassium it was 0.380  $\mu g/mL.$ 

## Limit of Quantitation (LOQ)

Based on the standard deviation of the response and the slope

The quantitation limit may be expressed as:

#### $LOQ = 10 \sigma / S$

For a tenolol LOQ was found to be 2.600  $\mu g/mL$  and for losartan potassium it was 1.160  $\mu g/mL.$ 

## Stability

Solution containing 10 µg/mL of the atenolol and 5 µg/mL losartan of Nebistar SA was analyzed by UV spectrophotometry method at 1, 2, 3, 5, 24, 48, 72, hours after preparation. The behavior of the analyte remained unchanged up to 4 days. All the measurements were made at room temperature (27-28°C). Tablet analysis % recovery was found to be 99.54  $\pm$  0.12 for AT and 99.32  $\pm$  0.08 for LS.

The results of validation are summarizes in Table 11.

Table 11:	Validation	data for	the deve	loped U	V		
spectroscopic method							

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Validation	Losar-Beta						
Parameters	AT	LS					
Linearity (r2)	0.9997	0.9995					
Analyst variation	0.1000	0.0700					
Inter day Variation	0.0200	0.0060					
Accuracy (%SD)	0.12	0.08					
(%found)	99.54	99.32					
LOD	0.860	0.380					
LOQ	2.600	1.160					

## **4.** CONCLUSION

The UV spectrophotometric method developed is simple, precise, rapid, selective and economical for the simultaneous estimation of atenolol and losartan potassium in solid dosage form. It can also be used for the analysis of these drugs in biological fluids and in quality control laboratories.

#### **REFERENCES**

- [1] Everett B. M., Robert J., Glynn, Eleanor Danielson, and Paul M Ridker, Combination therapy versus monotherapy as initial treatment for stage-2 hypertension, *Clinical Therapeutics* 30 (4), 2008, 661-671.
- [2] Fogary Roberto, Zoppi Annalisa, Mugellini Amedeo, Preti Paola Destro Maurizio, Rinaldi Andrea and Derosa Giuseppe; Effectiveness of hydrochlorothiazide in combination with telmisartan and olmesartan in adult with moderate hypertension not controlled with monotherapy, a prospective, randomized, openlevel, blinded end pont, parallel arm study, *Current Therapeutic research*; 69, I, 2008, 1-15.
- [3] Panderi I.E. Simultaneous determination of benazepril hydrochloride and hydrochlorothiazide in tablets by second-order derivative spectrophotometry, *J. Pharm. Biomed. Anal*, 21, 1999, 257-265.



- [4] Erk N. Analysis of binary mixtures of losartan potassium and hydrochlorothiazide by using high performance liquid chromatography, ratio derivative spectrophotometric and compensation technique, *J. Pharm. Biomed. Anal.* 24, 2001, 603–611.
- [5] Jeffery, G.H., Bassett, J., Mendham, J. and Denrey, R.C., In; "Vogel's Text book of Quantitative Chemical Analysis" 5<sup>th</sup> Edn., Longman Group U.K. Ltd., England, 1989, 3, 6-14.
- [6] Sethi, P.D., In; "Quantitative Analysis of Drugs in Pharmaceutical Formulations" 3<sup>rd</sup> Edn., CBS Publishers and Distributors, New Delhi, 1997, 9.
- [7] Beckett, A.H. and Stenlake, J.B. "Practical Pharmaceutical Chemistry" Vol. II, 4<sup>th</sup> Edn., CBS Publishers and Distributors, New Delhi, 1989, 276-99.

# **AUTHOR'S PROFILE**



Dr. Asmita Gajbhiye has about 15 years experience of research and teaching experience at both UG and PG levels. She is a well renowned scientist who has published more then 25 papers in journals of international and national repute and presented more then 50 papers in the various conferences/ seminars and symposia at national and international level. She has successfully completed the various research projects at PG and Ph. D. level. She has also received the best presentation awards at national level. Her research projects have been appreciated at international level during presentation of research papers. She has delivered invited lectures and chaired many sessions in several National and International conferences and symposia in India and abroad. Presently, she is working as Associate Professor in Department of Pharmaceutical Sciences, Dr. H.S. Gour Central University, Sagar, MP.